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□ In pharmacy this is always in a volume, normally expressed as mmol/mL, mmol/L, μmol/L or mEq/L □ A 1 molar solution contains 1 mole of material dissolved in 1 litre. □ Starting point it is possible to calculate the amount of material required for different volumes of different strengths of solution. 2014/06/03 Faculty of Pharmacy, Omer Al-Mukhtar University, Tobruk, Libya. 12

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Physicochemical principles of pharmacy Florence, A. T. (Alexander Taylor) ; Attwood, D This well-known textbook uses pharmaceutical examples to illustrate the underlying physical chemistry of drug delivery, characterisation, formulation, transport and absorption

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