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Physical chemistry for the life sciences

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Hydrogen bonds: The interaction between a hydrogen atom and one of the oxygen or nitrogen atoms forms hydrogen bonds. For example, hydrogen bonds in water can be shown as below: Hydrogen bonds play a major role in determining the shape of biological macromolecules such as proteins. Hydrogen bonds are 10 percent as strong as covalent bonds.

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3 Applications of Physical Chemistry in Daily Life - AZ ...

Free Download Physical Chemistry for the Life Sciences (second edition) written by Peter Atkins (Professor of Chemistry, Oxford University) and Julio de Paula (Professor of Chemistry, Lewis & Clark College) and published by W. H. Freeman and Company, New York in 2011.

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