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Electrochemical Engineering, Thomas F. Fuller and John N. Harb = 10–14 Chapter 2 Problem 2.19 1/1 Determine the solubility product K_{sp} for $PbSO_4$. The desired equilibrium is $PbSO_4 \leftrightarrow Pb^{2+} + SO_4^{2-}$. We can write this as the sum of two electrochemical equations $PbSO_4 + 2e^- \leftrightarrow Pb + SO_4^{2-}$ $Pb \leftrightarrow Pb^{2+} + 2e^-$ $\Delta G = \Delta G_1 + \Delta G_2 = 0$

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Electrochemical Engineering [Book]

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