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1.11 The force on a spring is described by:  $F = Kx$  where  $K$  is the spring constant. First calculate  $K$  based on the earth measurement then  $g_{Mars}$  based on spring measurement on Mars.

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Chemical Engineering Thermodynamics

$K_{\text{Wact}} = 125 \text{ mol}$  The solution is  $T_f = 549.39 \text{ K}$   
 $c_p = 29.1 \text{ J mol}^{-1} \text{ K}^{-1}$  solution The actual is work  $499.14 \text{ K}$ . is 25% Then greater  $h$  (b)  
Repeat the calculation with a temperature-dependent heat capacity?  
 $10^2 T$ ?  $3499$ ?  
 $10^5 T^2 + 7.464 \times 10^9 T^3$ . .  $C_p(T) = 22.243 + 5977$   
Assuming reversibility  $T_f = 479.44 \text{ K}$ .

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